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(CHLORODIFLUOROMETHYL)TRIFLUOROOXIRANE: SYNTHESIS AND REACTIONS

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The low-temperature nucleophilic reaction of 3-chloropenta-fluoropropene with alkaline hydrogen peroxide under the PTC conditions was studied. Instead of obvious ${\tt Cl}^-$ substitution, epoxidation took place to yield (chlorodifluoromethyl)trifluoro-oxirane.

$$[CF_2 = CF - CF_2 - OOH] \xrightarrow{OOH} - CC1F_2 - CF = CF_2 \xrightarrow{OOH} - CC1F_2 - CF - CF_2$$

The title epoxide was reacted with various nucleophiles and electrophiles. For example, the hydrate of chlorodifluoropyruvic acid was obtained in a good yield by the reaction with water in dioxane.

The action of the 1^{st} group metal fluorides or chlorofluoropropoxides on the title epoxide resulted in the formation of perhalogenated telomeric acyl fluorides.

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$$CC1F_2$$
 CF_2 + $R_{C1,F}$ CF_2 0 $R_{C1,F}$ $CC1_{D}$ 0 $R_{C1,F}$ $R_{$

The chemical behaviour of (chlorodifluoromethyl) trifluoro-oxirane was found to be similar to that of trifluoro(trifluoromethyl) oxirane and will be discussed further.